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#### Key indicators

Single-crystal X-ray study  
T = 294 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.012 \text{ \AA}$   
R factor = 0.069  
wR factor = 0.197  
Data-to-parameter ratio = 22.2

For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

## Bis( $\mu$ -3,5-diphenylpyrazolato- $\kappa^2\text{N}:\text{N}'$ )- bis[(4,4'-dimethyl-2,2'-bipyridine- $\kappa^2\text{N},\text{N}'$ )- palladium(II)] bis(hexafluorophosphate)

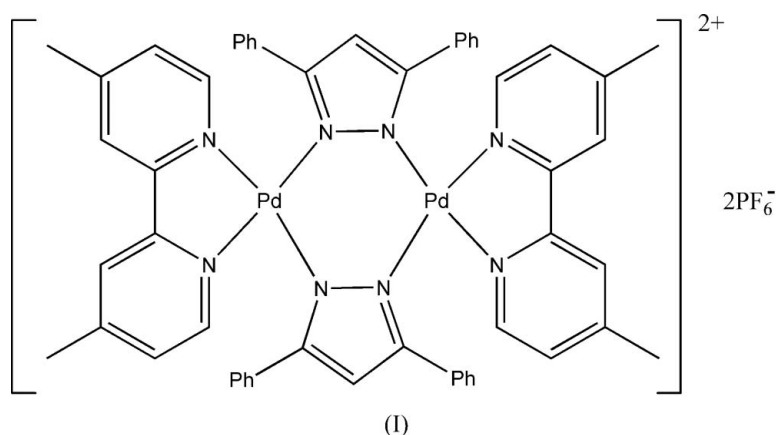
In the crystal structure of the title salt,  $[\text{Pd}_2(\text{C}_{15}\text{H}_{11}\text{N}_2)_2(\text{C}_{12}\text{H}_{12}\text{N}_2)_2](\text{PF}_6)_2$ , two (4,4'-dimethyl-2,2'-bipyridine)palladium(II) units are bridged by two pyrazolate ligands in an exodentate fashion. The six-membered ring consisting of the two Pd atoms and the four pyrazolyl N atoms has a boat-shaped conformation and the Pd atoms are four-coordinate in square-planar environments. The cation and both anions have crystallographically imposed mirror symmetry.

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### Comment

In previous publications, we have reported several pyrazolate-bridged dinuclear palladium(II) complexes having an inorganic anion as the charge-balancing species (Huang *et al.*, 2005), as well as a molecule that crystallizes as a solvate (Huang *et al.*, 2006). In the present paper, we report a dinuclear palladium(II) complex based on 4,4'-dimethyl-2,2'-bipyridine (dmbpy) and 3,5-diphenylpyrazolate ligands, *viz.* (I) (Fig. 1).



The  $\text{Pd}^{\text{II}}$  center has a *cis*-square-planar geometry defined by an *N,N'*-bidentate 3,5-diphenylpyrazolate anionic ligand and a chelating 4,4'-dimethyl-2,2'-bipyridine ligand. The dihedral angle between the two pyrazole planes is  $72.8(2)^\circ$ . In the cation a crystallographic mirror plane passes through C13, C21 and the mid-points of the N—N bonds. In one anion, P1, F1, F3, F4 and F5 lie on a mirror plane; in the other anion, P2, F7 and F8 lie on a mirror plane. The cations and anions are held together only by electrostatic interactions.

### Experimental

A mixture of (4,4'-dimethyl-2,2'-bipyridine)dinitratopalladium(II) (41.5 mg, 0.10 mmol) and 3,5-diphenylpyrazole (21.0 mg, 0.10 mmol) was dissolved in water (5 ml). To the mixture was added a tenfold

excess of potassium hexafluorophosphate, which resulted in the immediate deposition of yellow microcrystals. The crystals were filtered off, washed with a minimum amount of cold water and dried under vacuum (quantitative yield of 114.6 mg). Crystals were obtained by the vapor diffusion of diethyl ether into a 1 mM solution in acetonitrile.  $^1\text{H}$  NMR (400 MHz,  $[\text{D}_3]$  acetonitrile):  $\delta$  2.45 (12H, s, dmbpy-CH<sub>3</sub>), 7.17 (4H, d,  $J = 5.9$  Hz, dmbpy-H<sub>5,5'</sub>), 7.20 (2H, s, Ph<sub>2</sub>Pz-H<sub>4</sub>), 7.42 (12H, m, Ph-H) 7.78 (4H, d,  $J = 5.8$  Hz, dmbpy-H<sub>6,6'</sub>), 8.05 (4H, s, dmbpy-H<sub>3,3'</sub>), 8.26 (8H, m, Ph-H).

#### Crystal data

$[\text{Pd}_2(\text{C}_{15}\text{H}_{11}\text{N}_2)_2(\text{C}_{12}\text{H}_{12}\text{N}_2)_2](\text{PF}_6)_2$   
 $M_r = 1309.73$   
 Trigonal,  $R\bar{3}m$   
 $a = 38.2597$  (8) Å  
 $c = 12.3356$  (6) Å  
 $V = 15637.7$  (8) Å<sup>3</sup>  
 $Z = 9$

$D_x = 1.252$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 $\mu = 0.63$  mm<sup>-1</sup>  
 $T = 294$  (1) K  
 Prism, yellow  
 $0.35 \times 0.14 \times 0.14$  mm

#### Data collection

Rigaku Saturn70 diffractometer  
 $\omega$  scans  
 Absorption correction:  $\psi$  scan  
 (North *et al.*, 1968)  
 $T_{\min} = 0.89$ ,  $T_{\max} = 0.92$

40292 measured reflections  
 8178 independent reflections  
 7667 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.037$   
 $\theta_{\text{max}} = 27.4^\circ$

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.069$   
 $wR(F^2) = 0.197$   
 $S = 1.08$   
 8178 reflections  
 369 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.154P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 1.09$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -1.44$  e Å<sup>-3</sup>  
 Absolute structure: Flack (1983),  
 with 4023 Friedel pairs  
 Flack parameter: 0.02 (4)

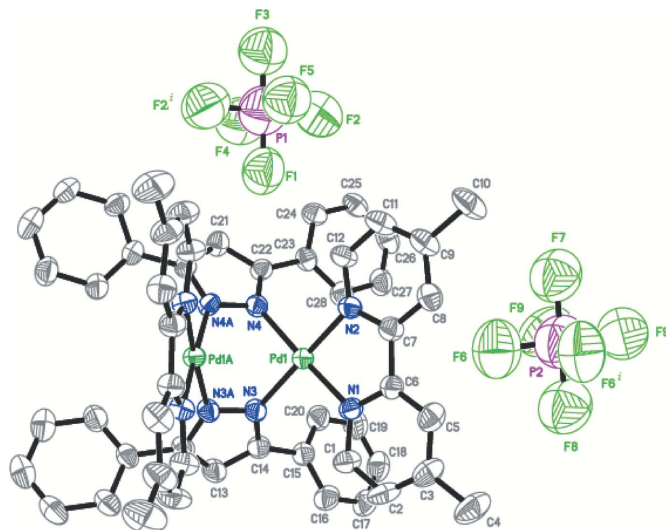
**Table 1**

Selected geometric parameters (Å, °).

Pd1—N1	2.025 (5)	Pd1—N4	2.042 (5)
Pd1—N2	2.016 (5)	Pd1—Pd1 <sup>i</sup>	3.0410 (8)
Pd1—N3	2.040 (5)		
N1—Pd1—N2	81.03 (19)	N2—Pd1—N3	175.8 (2)
N1—Pd1—N3	96.7 (2)	N2—Pd1—N4	95.0 (2)
N1—Pd1—N4	175.3 (2)	N3—Pd1—N4	87.0 (2)

Symmetry code: (i)  $x, x - y, z$ .

Aromatic H atoms were constrained to an ideal geometry, with C—H distances of 0.93 Å and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . Methyl H atoms were rotated to fit the electron density, with C—H distances of 0.96 Å and with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ . The two hexafluorophosphate



**Figure 1**

The structure of the constituent ions of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme. In the cation, symmetry code: (A)  $x, x - y, z$ . In the P1-containing anion, symmetry code: (i)  $x, x - y, z$ . In the P2-containing anion, symmetry code: (i)  $1 - x + y, y, z$ .

anions are disordered. The F—P distances were restrained to 1.58 (1) Å and F...F distances to 2.23 (1) Å. The largest peak and deepest hole in the final difference Fourier map are located 4.31 and 0.12 Å from atoms H10C and P2, respectively.

Data collection: *CrystalClear* (Rigaku/MSK, 2004); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

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